Quantification of Oleanolic acid in the flower of *Gentiana olivieri* Griseb. by HPLC

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ABSTRACT

High Performance Liquid Chromatography (HPLC) method is a potent analytical technique for determining phytoconstituents even in minute quantities particularly that are generally present in traces concentrations in medicinal plants. The concentration of oleanolic acid (OA) in the *Gentiana olivieri* Griseb. (GOG) flowers is quantitatively determined by a simple, precise and accurate HPLC method. HPLC assay was performed on reversed-phase C18 column and compound was detected at 210 nm with a flow rate of 1.0 ml/min. The mobile phase consisted of methanol (A) and 0.03 mol/L phosphate buffer, pH 2.9 (B) with a ratio of 85:15 (A: B v/v). The method showed good precision and accuracy with overall intra-day and inter-day variation of 0.36-1.33% and 0.43-1.15%, respectively, and overall recoveries of 99.32%. Application of these methods to determine the OA in *Gentiana olivieri* Griseb showed that the concentration of OA was more in the flower as compared to the whole aerial part (average amount in flower 1.82 mg/g DW and 0.75 mg/g DW in whole aerial plant). A simple and accurate HPLC method has been developed for quantitative determination of OA in GOG flowers.

INTRODUCTION

*Gentiana olivieri* Griseb. (GOG) is a perennial herb found in temperate region of Gulf countries, middle Asia, some temperate region of China, western Himalayas and Pakistan[1]. It is traditionally used in Ayurveda and Unani systems of medicines in arthritis, anti-inflammatory, antidepressant, antiulcerogenic, gastroprotect and sudorific[2-3]. In Turkey the drug is widely used as bitter tonic, appetizer and in some mental disorders[4]. The plant has reported many pharmacological activities as hypoglycaemic, anti-hypertensive[5] and hepatoprotective[6]. Furthermore, several alkaloids, fatty acids, secoiridoids, triterpenoids (oleanolic acid (OA) and ursolic acid from flowers) and bioflavonoids were isolated from the plant[7]. Oleanolic acid is ubiquitous triterpenoid in plant kingdom, medicinal herbs, and is integral part of the human diet. During the last decade over 300 research articles have been published on the research, reflecting tremendous interest has prompted us to develop quantitative HPLC analytical method in the flower of Gentiana olivieri Griseb[8].

OA is a main triterpenic acid reported in GOG flowers. OA has proved to have pharmacological activities such as anti-inflammatory[9], diuretic, anti-tumor[10], and Anti-HIV [11-12]. It has also been reported as anti microbial[13], anti fungal[14], gastroprotective[15] and hypoglycemic[16] properties. It is relatively nontoxic and has been used in number of health products. The present research on this plant involving extraction, identification and pharmacological studies mainly focused on the entire plant with little or no work on flowers of plant which are important source of oleanolic acid. No HPLC method has been reported earlier on the quantification of OA in the flowers of GOG. Henceforth we are reporting in this communication development of simple and accurate HPLC method for determination and quantification of OA in GOG flowers.

MATERIALS AND METHODS

The HPLC studies on GOG for determination of oleanolic acid was carried in October 2010 at ASBASJSM College of Pharmacy Research Laboratory, Bela, Ropar (India).

**Chemicals**

OA was purchased from Sigma Chemical Co. (India). Methanol and water used were of HPLC grade. All other chemicals used were of analytical grade.

**Preparation of standard solution**

A stock solution OA, 1 mg/ml was prepared in methanol. A serial dilution were made on the stock solution with methanol to prepare the concentration of 50, 100, 200, 300 and 400 μg/ml, from each of which 20 μl was used to plot standard curve for OA.

**Chromatographic apparatus and conditions**

The assay method for quantitative determination of OA in GOG flowers was based on method developed by Chunhua et al., 2007[17] with
modifications. Quantification of OA was performed on an Ultra Performance Liquid Chromatography (UPLC) system (Shimadzu Japan) equipped with LC20A pump, UV detector LC20AD and ODS C18 column (250 × 4.6 mm, 5μm). This compound was detected at 210 nm at 20°C with an eluent flow rate of 1 mL/min. The mobile phase consisted of methanol (A) and 0.03 mol/L phosphate buffer (pH 2.9) (B) with a ratio of 85:15 for detection of OA.

Collection and preparation of plant material
GOG herb was purchased from Himalaya Herbal Store, Saharanpur (UP), India and got authenticated at National Institute of Science Communication and Information Resources, New Delhi (the sample of which is deposited with this institute) wide letter No. NISCAIR/RHMD/Consult/2009-10/1255/59. The plant was dried, the flower and aerial part separated, grounded into powder. A plant powder aliquot of 0.5g was solubilized in 20 ml ethanol for 2 h followed by 30 min ultrasonic extraction. The samples were extracted twice and both extracts were combined and evaporated to dryness below 40°C. The residue was dissolved in 1 ml methanol filtered through 0.4μm millipore before HPLC analysis.

Validation of the assay
The linearity of the method for each standard was established by injection of 20 μL of standard solutions of different concentrations with five replicates. Calibration graphs were plotted subsequently for linear regression analysis of the peak area with different concentrations i.e. 50, 100, 200, 300, 400 μg/mL at 210 nm absorbance (Fig. 1). To determine the intra-day precision for each standard, five injections of each concentration were done on the same day. These studies were repeated on three consecutive days to determine the inter-day precision. To determine the extraction recovery, dried flower samples was added with 0.20 mg OA, standard before extraction. The follow-up extractions and HPLC analyses were operated in the same manner as above. The recoveries were calculated as follows:

\[ \text{Recovery (\%)} = \left( \frac{A - B}{C} \right) \times 100\% \]

where A is the amount detected, B is the amount of sample without standards and C is the spiked amount of the standards.

RESULTS AND DISCUSSION
Identification of OA by HPLC
OA, standards were all detectable under a wavelength of 210 nm. OA could be eluted efficiently and detected by a mobile phase consisting of methanol (A) and 0.03 mol/L phosphate buffer (pH 2.9) (B) with a ratio of 85:15 (A:B, v/v). With a flow rate of 1.0 mL/min, the mean retention time was 15.43 as shown in Fig. 1, Fig. 2(a), Fig. 2(b) and Fig. 2(c). The linearity of calibration curve for OA is validated by the high value of correlation coefficient (i.e. 0.9995). The concentrations were determined using calibration standard curves prepared for OA (Fig. 1). Fig 2a, 2b, and 2c are the respective HPLC chromatograms of standard, flower and aerial part, the results have shown good accuracy and precision (NMT 4.09%) obtained over the entire selected range of concentrations. The intra- and inter-day variability studied over three consecutive days was NMT 1.33% which is statistically considered to be significant.

Validation of the developed method
Linearity
Linearity studies were carried out to determine whether the established methods could measure different concentrations of OA accurately. OA was determined by measuring the peak area. Five different concentrations, i.e. 50, 100, 200, 300 and 400 μg/mL were made for the standard and tested respectively with five repetitions for each concentration. A plot of mean peak area against concentrations of OA gave a linear relationship (with a correlation coeffecient, \( r^2 = 0.9995, n=5 \)) over the concentration range of 50-400 μg/mL. The calibration curves between conc. of OA (x-axis) and peak area (y-axis) at 210 nm were all linear and the regression equation between the mean peak areas (y) against concentration (x) was:

\[ y = 10578x + 30957 \] (Fig 1)

Reproducibility and accuracy
Reproducibility and accuracy were determined for the standard sample at five different concentrations each as described above (Table 1). The intra-day coefficient of variation on five replica of each concentration was 0.36-1.33% and inter-day coefficient calculated over three consecutive days were 0.43-1.15%. Percentage accuracy for intra and inter-day were -3.96 to 2.8% and -4.09 to 2.71 respectively, showing good accuracy and reproducibility.
Quantiﬁcation of Oleanolic acid in the ﬂower of Gentiana olivieri Griseb by HPLC 

Extraction recoveries

To determine extraction recoveries, a measured amount of 0.20 mg OA, was added to the Gentiana olivieri Griseb ﬂower and analyzed as described above (n = 3). The ratio of the assay value to the spiked amount for the standards was calculated. The recovery of OA was 99.32 % data shown in (Table 2).

DISCUSSION

OA is an important constituent with a good food value and a number of pharmacological properties. As per the literature review no work has ever been reported for quantification of OA in GOG by HPLC method. OA was identiﬁed and subsequently determined by plotting linear graph with serial standard dilutions and extraction and preparation of plant extract. A good linearity with correlation coefﬁcient r², 0.9995 over the tested range of 50–400μg/mL for OA was obtained. A good analytical method should have a good reproducibility and accuracy. This method showed good reproducibility when tested over three consecutive days with intra and inter-day coefﬁcients of variation of 0.36–1.33% and 0.43–1.15% respectively showing clearly that variations due to time within the assay are negligible. The method showed good precision and accuracy with overall recoveries of 99.32%. Extraction recovery percentages from 80 to 120[18] are considered to be acceptable by international validation protocols. The recovery of an analyte need not be 100% but in extent of recovering of an analyte and of internal standard should be consistent, precise and reproducible[19]. The results showed that ﬂower contained the higher contents of OA as compared to whole aerial plant (average amount in ﬂower 1.82 mg/g DW and 0.75 mg/g DW in whole aerial plant). However, each plant part needs to be screened for the presence of OA. The results obtained for the determination of OA in Gentiana olivieri Griseb. supports the earlier work by Chunhua et al.,2007 and Shuge et al., 2010[17,20] to quantify OA by different methods on different plants with respect to its simplicity, quickness and accuracy. The results are encouraging enough to explore medicinal value of this plant based on this important bioactive constituent.

Figure 2: HPLC chromatogram of OA, (a) standard, (b) the ﬂower extract of Gentiana olivieri Griseb. (GOG) and (c) Aerial part extract of GOG.
CONCLUSION
In this study, simple and accurate HPLC method for the quantification of OA in the flowers of plant *Gentiana olivieri* Griseb. was developed and validated. The present assay method is the first to be reported for the determination of OA in GOG plant. The whole assay can be completed within about one hour. Another advantage is that minimum sample volume (20μL) is required.

It was concluded that the concentration of OA was found more in the flower (1.82mg) as compared to the whole aerial part (0.75mg). Such detection systems can be used to quantify this bioactive component in *Gentiana olivieri* Griseb and other related medicinal plants in the future. This method can be a useful tool for quality control of pharmaceutical formulations.

ACKNOWLEDGMENT
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CONFLICT OF INTEREST
Authors declare no conflict of interest

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**REFERENCES**


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**Table 1:** Reproducibility and accuracy analysis of Oleanolic acid (*n* = 5).

<table>
<thead>
<tr>
<th>Standard</th>
<th>Nominal concentration (μg/mL)</th>
<th>Assay value (mean ±SD) (μg/mL)</th>
<th>Coefficient of variation (%)</th>
<th>Accuracy (%)</th>
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<tbody>
<tr>
<td>Intra-daya</td>
<td>50</td>
<td>48.02 ± 0.64</td>
<td>1.33</td>
<td>-3.96</td>
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<td></td>
<td>100</td>
<td>99.29 ± 0.87</td>
<td>0.87</td>
<td>-0.71</td>
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<tr>
<td></td>
<td>200</td>
<td>205.61 ± 0.74</td>
<td>0.36</td>
<td>2.8</td>
</tr>
<tr>
<td></td>
<td>300</td>
<td>298.29 ± 3.44</td>
<td>1.15</td>
<td>-0.57</td>
</tr>
<tr>
<td></td>
<td>400</td>
<td>399.25 ± 2.36</td>
<td>0.59</td>
<td>-0.19</td>
</tr>
<tr>
<td>Inter-dayb</td>
<td>50</td>
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<td>0.92</td>
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<tr>
<td></td>
<td>100</td>
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</tr>
<tr>
<td></td>
<td>200</td>
<td>205.43 ± 0.89</td>
<td>0.43</td>
<td>2.71</td>
</tr>
<tr>
<td></td>
<td>300</td>
<td>298.27 ± 3.43</td>
<td>1.15</td>
<td>-0.58</td>
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<tr>
<td></td>
<td>400</td>
<td>399.2 ± 2.37</td>
<td>0.59</td>
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</tr>
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</table>

*aThe sample was analyzed five times within one day.
*bThe sample was analyzed over three consecutive days.

**Table 2:** Recovery tests of the analytic methods for Oleanolic acid (*n* = 3).

<table>
<thead>
<tr>
<th>Standard</th>
<th>Spiked amount</th>
<th>Assay value (mean ±SD) (μg/mL)</th>
<th>Recovery rate (%)</th>
</tr>
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<tbody>
<tr>
<td>Oleanolic acid</td>
<td>0.20</td>
<td>0.199 ± 0.007</td>
<td>99.32 ± 3.50</td>
</tr>
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</table>